Self-Oxidation of Reserve Proteins Enriched With Hydrogen SOV/20-120-3-39/67

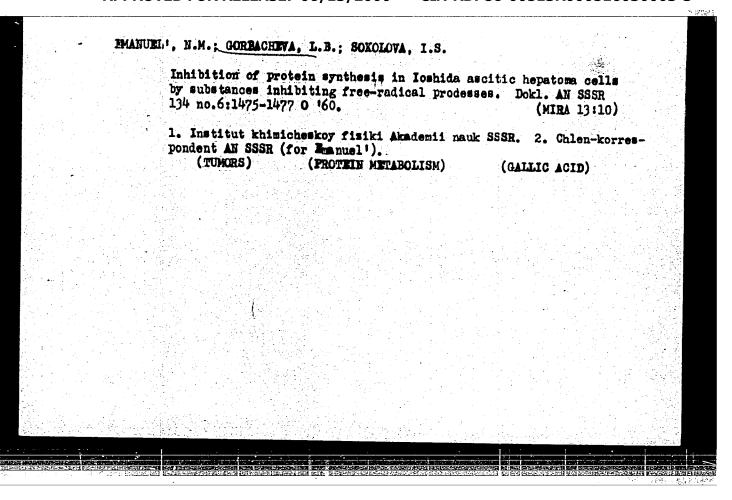
They showed a different degree of reduction and were stored for different periods. A storing in an air-dried state reduced the reduction power of legumin, which apparently was gradually oxidized by atmospheric oxygen. Table 2 shows results of the experiments dealing with the oxidation velocity of glycinin. The authors are of opinion that the oxidation by K.Fe(CN) is accompanied by the same effects of atmospheric oxygen, which fact is proved by Table 3. The movable oxygen of organic substances (N-oxidized alkaline form) has the same effect. The possibility of the formation of peroxides can be assumed as explanation (Ref 3). The conducted experiments proved this assumption. Tables 4 and 5 show, however, that glycin and edestin consume smaller quantities of H₂O₂ than the initial proteins, when they are reduced. This phenomenon has hitherto not been cleared up. Nevertheless it is possible to draw the conclusion that the reserve proteins of the seeds enriched with oxygen can take part in several metabolic processes in the cell, among them in respiration, by means of binding the free oxygen. There are 5 tables and 6 references, 6 of which are Soviet.

Card 2/3

Self-Oxidation of Reserve Proteins Enriched With Hydrogen SOV/20-120-3-39/67
PRESENTED: January 18, 1958, by A. I. Oparin, Member, Academy of Sciences, USSR
SURMITTED: January 18, 1958

1. Proteins--Oxidation 2. Proteins--Physiological effects 3. Proteins--Properties

Card 3/3



CORBACHEVA, L.B., EMANUEL, N.M. (USSR)

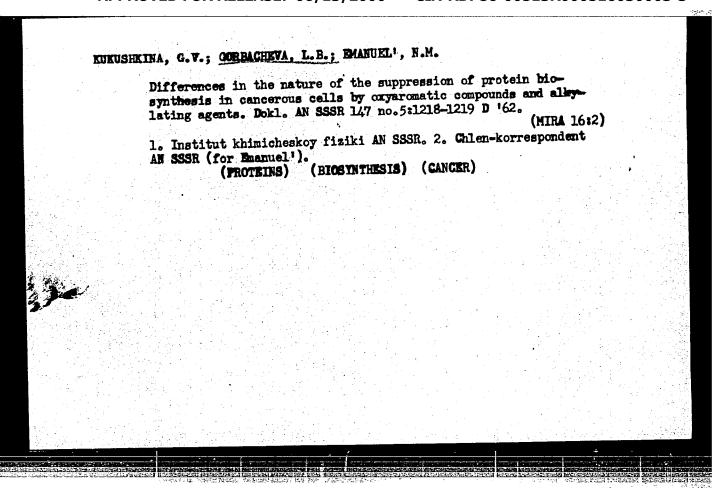
"Suppression of Cell Protein Biosynthesis in Yoshida Ascites Hepatoma by Free-Radical Reaction Inhibitors."

Report presented by the 5th Biochemistry Congress "oscow, 10-16 Aug 1961.

KUKUSHKINA, G.V.; GORRACHEVA, L.B.; EMANUEL¹, N.M.

Kinetic characteristics of the inhibition of protein biosynthesis in cancer cells treated with alkyl phenols and chlorampenicol. Dokl. AN SSSR 146 no.5:1206-1208 0 '62. (MIRA 15:10)

1. Institut khimicheskoy fiziki AN SSSR. 2. Chlen-korrespondent AN SSSR (for Reamel¹). (CHLORAMPHENICOL—PRISIOLALICAL EFFECT) (PHENOLS—PHYSIOLOGICAL EFFECT)



ACC NR: AP6032114 SOURCE CODE: UR/0301/66/012/005/0452/0455 AUTHOR: Kukushkina, G. V.; Corbacheva, L. B.; Emanuel', N. M. ORG: Institute of Chemical Physics, Academy of Sciences SSSR (Institut khimicheskoy fiziki AN SSSR) TITLE: Inhibition of the biosynthesis of protein and nucleic acids by phenolic compounds in vivo SOURCE: Voprosy meditsinskoy khimii, v. 12, no. 5, 1966, 452-455 TOPIC TAGS: biochemistry, biosynthesis, protein synthesis, nucleic acid, and phenol derivative, metabolic effect, ionole ABSTRACT: In vivo experiments on mice affected with Erlich ascites tumor and hepatoma XXII showed that the phenol derivatives propylgallate and ionole (4-methyl-2,6-di-tert-butyl-phenol) inhibited protein and nucleic acid synthesis in some organs and tissues. Propylgallate did not affect protein biosynthesis in normal kidney tissue but was effective against cancerous tissue. A 200 mg/kg dose of ionole suppressed uptake of C14 labeled amino acids almost completely. Further experiments showed that the cellular nucleic acid fractions from cancerous cells were the most sensitive to the action of these compounds. [WA-50; CBE No. 12] 06/ SUBM DATE: 19Nov64/ ORIG REF: 014/ OTH REF: 002/ UDC: 615.778.1-092:612.015.348-064+616.015.348.014. SUB CODE:

Zhur.ob.khin	1. 33 no.2:511-515	of halogen in a pyr. F 163.	excile ring. (MIRA 16:2)	
l. Moskovski	y gosudarstvennyy (Pyrazole)	universitet imeni (Halogens)	M.V.Lomonosova.	

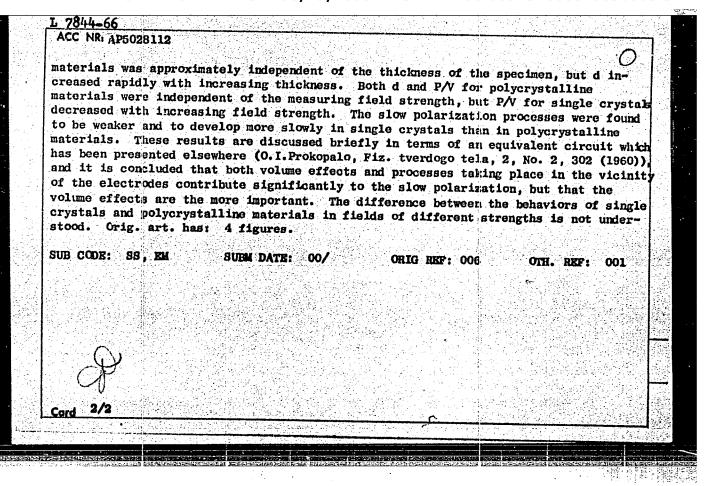
GRANDHERG, I.I.; GORBACHEVA, L.I.; KOST, A.N.; SIBIRYAKOVA-FEDOTOVA, D.V.

Pyrazoles. Part 33: Oxidative elimination of a bensyl group and interaction of the Gringmard reagent with halopyrasoles. Zhur.ob.khim. 33 no.2:515-519 F '63. (MIRA 16:2)

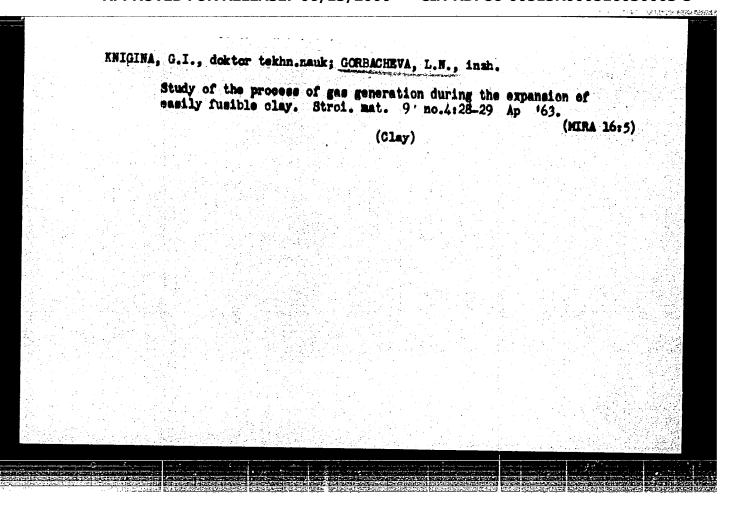
1. Moskovskiy gosudarstvennyy universitet imeni M.V.Lomonosova. (Benzyl group) (Grignard reagents) (Pyrazole)

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salah di kacamatan	1. Mosko	vskiy gos	udars tvenny	y universit	et imeni l	M.V.Lomo		

EWP(a)/EPA(a)-2/EWT(m)/EWP(i)/EPA(w)-2/EWP(t)/EUP(b) 28112 ID/WH SOURCE CODE: UR/0048/65/029/011/2026/2028 <u>l 7844-66</u> ACC NR. APIO28112 JD/WH AUTHOR: Gorbacheva, L. K .: Prokopalo, O.I. ORG: Rostov-on-the Don State University (Rostovskiy-na-Donu gosudarstvennyy universitet) TITLE: Mechanism of slow polarization processes in barium titanate Report, Fourth All-Union Conference on Ferro-electricity held at Rostov-on-the Don 12-16 September D 464 SCURCE: AN SSSR. Izvestiya. Seriya fizicheskaya, v. 29, no. 11, 1965, 2026-2028 TOPIC TAGS: ferroelectric material, barium titanate, single crystal, polycrystal, dielectric constant, electric polarization ABSTRACT: The authors have measured the ratio P/V of the slow polarization emf to the polarizing voltage and the dc dielectric constant d of a number of barium titanate single crystals and polycrystalline specimens which were subjected to a polarizing field and subsequently short circuited for definite lengths of time. The experimental technique is described elsewhere (O.I. Prokopalo, Sb. Sagnetoelektriki, str. 112, Izv. Rostovsk. un-ta, 1961). The polycrystalline specimens were from 1 to 8 mm thick and were prepared by the usual technique; the single crystals were 0.5 mm thick and were grown from a welt in KF. Fired on silver electrodes were employed and the measuring field was varied from 10 to 80 V/cm. The ratio P// for polycrystalline Card 1/2



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	Mechanism titanate.	underlying slow polarization processes in barium Izv. AN SSSR. Ser. fiz. 29 no.11:2026-2028 N '6. (MIRA 18:11)	
	1. Rostovi	kiy-na-Donu gosudarstvennyy universitet.	



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					(MIRA	17:2)	

TEPLITSKAYA, Ye.S.; MALAYA, L.P.; MIRGORODSKAYA, A.K.; SHEYKO, Z.A.;
KOGAN, TS.I.; OSIFUVA, Ye.S.; GIRGORASH, N.G.; PANKRATOVA, V.S.;
GORBACHEVA, L.Ye.

Species of dysentery pathogens encountered in 1959 in certain regions of Dnepropetrovsk Province and their sensitivity to the dysentery bacteriophage and antibiotics. Vrach. delo no.9:116-118 S '61.

(DNEPROPETROVSK PROVINCE—SHIGELLA)

(BACTERIOPHAGE)

(ANTIBIOTICS)

Hygienic evaluation of storage conditions and quality of breast milk at a donor center [with summary in English]. Pediatriia 36 no.10:14-20 0 58 (NIRA 11:11)

1. Is sanitarno-epidemiologicheskoy stantsii Dsermhinskogo rayona Leningrada.

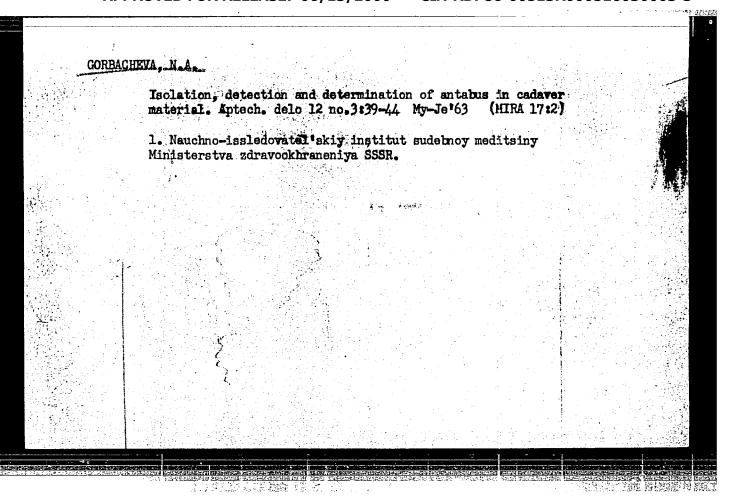
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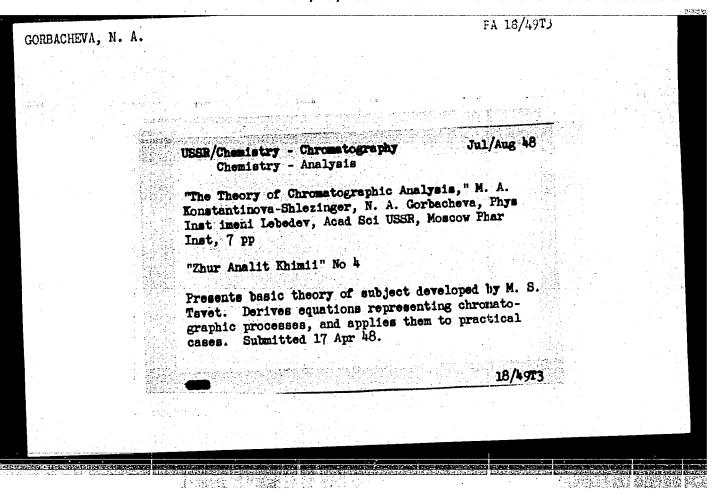
donor centers, determ, of milk quality & hyg.

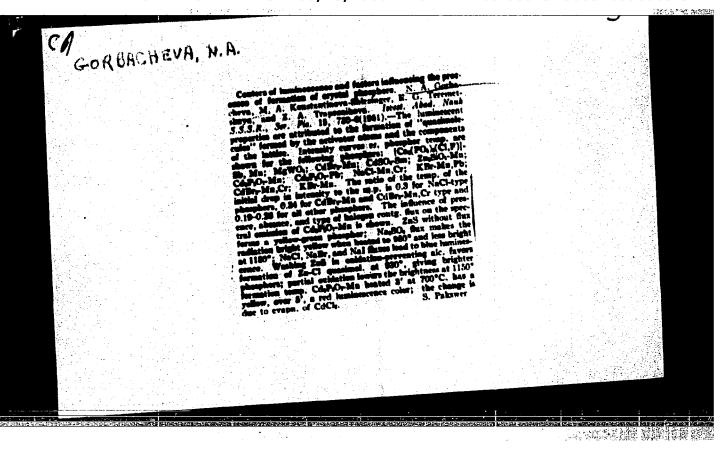
evaluation of storage cond. (Rus))

BABIN, V.B.; KOFMAN, I.L.; MANEVICH, A.Z.; MIKHEL'SON, V.A.; GORBACHEVA, M.P.;
YUREVICH, V.M.

Comparative evaluation of ether concentration in the blood in pure and in combined ether-oxygen anesthesia. Trudy 1-go MMI 33:324-332
(MIRA 18:3)





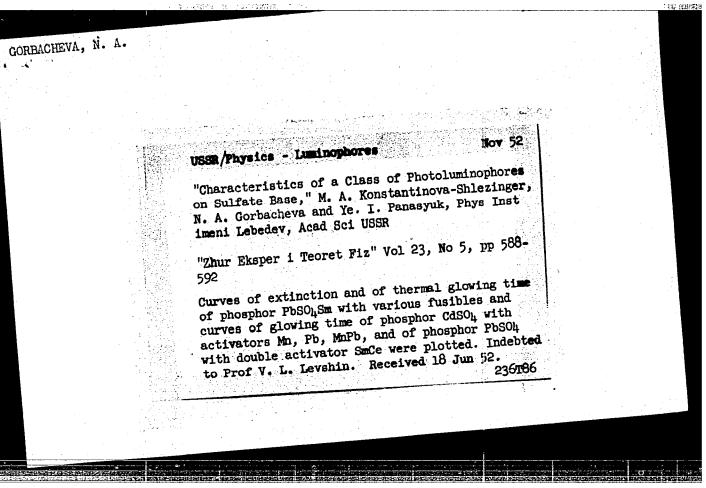


c A

Particularities of the laminessence of the Cd.P.O. Mn. Pb phosphor. N. A. Gorbacheva (P.N. Lebedev Phys. Inst. Acnd. Sci. U.B.S.R., Moscow). Zhur. Estpll. Teoret. Fis. 21, 305-19(1951).—The phosphor was made by fusion of 1 part Cdi.P.O. A × 10-2 Pt(NO2), 3 × 10-2 Mn phosphor phate, 2 × 10-2 CdFs, in the absence of Cl. and SO, ", and heating 30 min. at 000°; it contains a slight excess of P over the formula. In excitation with 254 ma, the emission is initially orange-rose, then veers to white. If, after a 1st evolution, the phosphor is left in the dark for sometime, and then excited again, one observes an initial reddish flash of an intensity substantially greater than the stationary brightness. Cdi.P.O. activated with Mn alone shows only reddish, and with Pb alone, only white-bhish laminescence. The observations are consistent with Antonov-Romanov-skil's ductrine of emission under the very action of the exciting light which lifts electrons from certain local levels and thus leaves them unfilled; on interruption of the evicitation these levels become occupied by electrons thermally liberated from other levels. On 2nd excitation, these trapped electrons are again liberated, thus giving rise to a flash; the necessary condition is that the phosphor be left in the dark between the 2 excitations. In expts. with a 1st 2-min. excitation, followed by a 8-min. dark period, with sep. measurements of the "Mn emission band" (green filter), the brightness of the Mn-band flash on 2nd excitation was found to be greater the more intense the 1st excitation, and excitation trausits only in normal growth of the brightness. This is taken as evidence that the Mn-band flash on 2nd excitation is brought under these conditions, 3nd excitation is brough to both the dark between the 1st excitations. This is taken as evidence that the Mn-band flash on 2nd excitation is brought of the should be the excitation and the production of the should be the excitation of the order of light liveli. The Pb band shows no flash, and decays faster than

the Ph band lines all the light stored, whereas on heating only the Mn band is emitted. However, the curves of growth of the brightness of the Pb band in 1st and in 2nd excitation do not coincide; at the initial stages, the intensity is distinctly higher on 2nd excitation. This means that the growth of the Pb band is influenced by the energy stored in the Mn band is not a consistency of the Pb band is influenced by the energy stored in the Mn band is not a consistency in the mn and in the Mn band is not a consistency in the mn and in the Mn band in the consistency in the mn and in the mn and in the mn and in the curve of the particular the mn and in t the Mn band, or, more accurately, depends on the no. of

electrons still remaining in deep levels after the 5-min, pros-phorescence. The cause of the different behaviors of the Ph and Mn bands flors not follow from the simple rephorescence. The cause of the different behaviors of the Ph and Mn bands does not follow from the simple recombination mechanism; there must be an added factor, specifically unimal, quenching in the Pb centers. This is borne out by the near-exponential character of the decay of the Pb band. Inasmuch as the Mn band suffers no quenching, its law of decay should be essentially hyperbolic, and this is confirmed. The singly activated Cd₂P₂O₂-Pb phosphor decays hyperbolically, I = (a + M) *, with a somewhat * I, the same in both the red- and the green filtered band. Consequently, in contrast to its behavior in the doubly activated phosphor suffers no unimal, quenching. The singly activated Cd₂P₂O₂-Mn phosphor also differs from the doubly activated Cd₂P₂O₂-Mn phosphor also differs from the doubly activated Cd₂P₂O₂-Mn, Pb by the absence of a flash on 2nd excitation. This shows that in Cd₂P₂O₂-Mn, Pb the flash is detd, by an interaction between the 2 activators; the quenching of the Pb band must be due to Mn, and the deep levels responsible for the flash depend on the simultaneous presence of both Mn and Pb. This is corroborated by the curves of thermally induced phosphoreacence of the 2 singly and the doubly activated phosphora. The Pb phosphor has a low peak at about 200°, the Mn Pb phosphor the highest peak at about 200°, the Mn Pb phosphor the highest peak at about 200°, the Mn Pb phosphor the highest peak at about 200°, the Mn Pb phosphor the highest peak at about 200°, the Mn Pb phosphor cannot be represented as any linear combination of the levels of the 2 singly activated phosphors.



CORBACHEVA

48-5-22/56

SUBJECT:

USSR/Luminescence

AUTHOR:

Gorbacheva N.A.

TITLE:

Photoluminopheres Based on Phosphates (Fotolyuminofory na

osnove fosfatov)

PERIODICAL: Izvestiya Akademii Nauk SSSR, Seriya Fizicheskaya, 1957.

Vol 21, #5, pp 682-683 (USSR).

ABSTRACT:

Over 40 phosphate luminophores have been produced. Luminophores synthesized of cadmium phosphates, which were obtained at different pH of the final solution, vary considerably in

luminosity and decay time.

The concentration of an activator affects the luminescence spectrum. It was established that the most luminous luminophores are produced by using as a base the cadmium phosphate obtained at room temperature from a solution whose pH-value

WAS 5.

The investigation allowed to recommend the following phosphate luminophores for luminescence tubes: barium-titanium-phosphate, whose spectrum is equal to magnesium tungstate; calcium ortho-

Card 1/2

48-5-22/56

TITLE:

Photoluminophores Based on Phosphates (Potolyuminofory na osnove fosfatov)

phosphate with cerium and manganese as activators, and calcium orthophosphate with tin and manganese, as activators, which can be used in luminescent tubes to make up deficient emission in the red region of spectrum.

There is a ground to assume that phosphate luminophores will take one of the first places among the tube luminophores.

The report was followed by a short discussion.

One Russian reference is cited.

INSTITUTION: Physical Institute in. Lebedev of the USSR Academy of Sciences

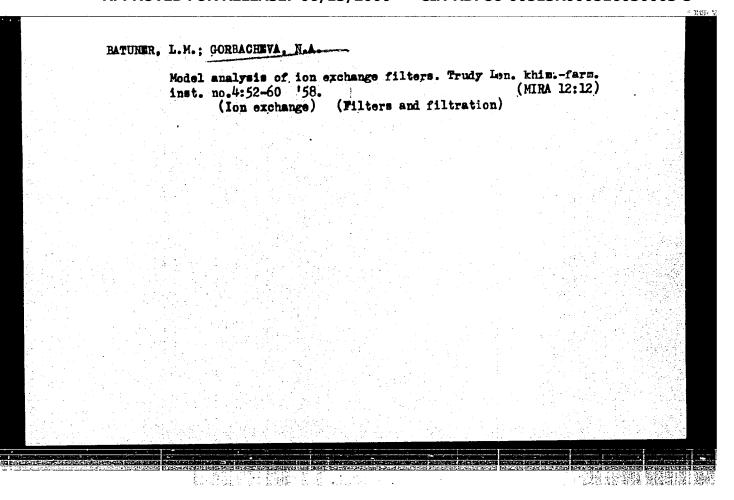
PRESENTED BY:

SUBMITTED: No date indicated

AVAILABLE: At the Library of Congress.

Card 2/2

GORBACHEVA, N.A.: Master Chem Sci (diss) -- "Photoluminophores based on the phosphates of cadmium". Moscow, 1958. 10 pp (Acad Sci USSR, Inst of Gen and Inorganic Chem im N.S. Kurnakov), 150 copies (KL, Nc 1, 1959, 114)



Use of trailon B in determining sinc in biological material. Apt. delo ? no.1:25-28 Ja-7 '58. (MIRA 11:2) l. Is kafedry sudebncy khimii (rukovoditel' - prof. M.D.Shvaykova) Moskovskogo farmatsevticheskogo institute. (AGNTIC ACID) (ZINC)	د. میراند میراند	GORBACHEVA, N.A., aspirant	
		Use of trailon B in determining sinc in biological material. Apt. delo 7 no.1:25-28 Ja-F '58. (MIRA 11:2)	
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Chromatographic separation of zinc from biological mandical investigations. Report no.1: separation of zinc elements by means of anion exchange chromatography. 3 no.4:35-40 O-D *60.	inc from some
1. Kafedra sudebnoy khimii (zav prof. M.D.Shvaykova ordena Lenina meditsinskogo instituta imeni I.M.Secher (ZINC) (CHROMATOGRAPHIC ANALYSIS)	a) I Moskovskogo nova.

BELOVA, A.V.; GORBACHEVA, N.A.; SHVAYKOVA, Mariya Dmitriyevna, prof.; SHEVERDYAYEVA, V.M.; RUBTSOV, A.F., kand.farmatsevticheskikh nauk, retsenzent; YASKINA, D.Z., kand.farmatsevticheskikh nauk, retsenzent; KOZULIN, V.S., red.; RAYKO, N.Yu., tekhn.red.

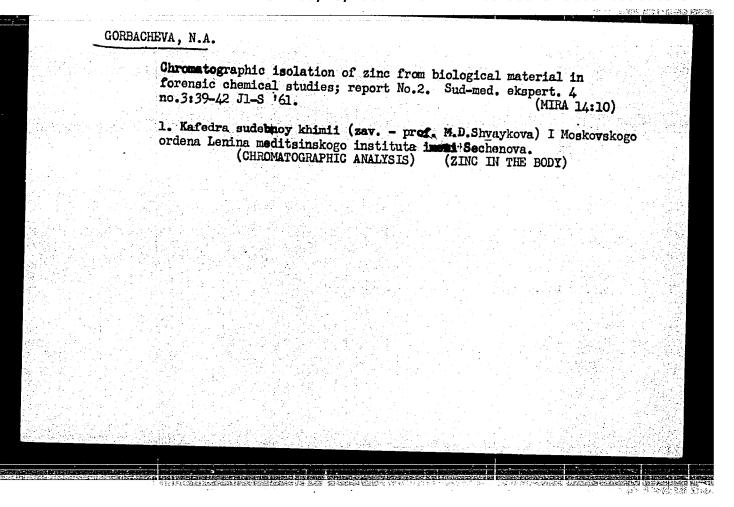
[Manual on the practical studies of forensic chemistry for

[Manual on the practical studies of forensic chemistry for pharmacology correspondence students of institutions of higher learning] Rukovodstvo k prakticheskim zaniatiiam po sudebnoi khimii; dlia studentov-zaochnikov farmatsevticheskikh vuzov.

Pod obshchei red. M.D.Shvaikovoi. Moskva, I-i Mosk.med.in-t im. I.M.Sechenova, 1961. 101 p.

(MIRA 14:6)

l. Kafedra sudebnoy khimii farmatsevticheskogo fakul'teta L-go Moskovskogo ordena Lenina meditsinskogo instituta imeni I.M. Sechenova (for Belova, Gorbacheva, Shraykova, Sheverdyayeva). (PHARMACOLOGY—LABORATORY MANUALS) (CHEMISTRY, LEGAL)



22155

S/048/61/025/004/004/048 B104/B201

94,3500

Gorbacheva, N. A. and Osiko, V. V.

TITLE:

AUTHORS:

Valence of Sn and Mn activators in crystal phosphors

PERIODICAL:

Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, v. 25,

no. 4, 1961, 454-455

TEXT: The present paper has been read at the 9th Conference on Luminescence (Crystal Phosphors), Kiyev, June 20-25, 1960. The authors have studied the relationship between the luminescence properties of Sn-and Mn-activated luminophores and the valence of these activators. The mean valences of Sn have been determined polarographically, and those of Mn by colorimetric and iodometric methods of analysis. Detailed results on Mn-activated luminophores have been earlier published by Osiko et al. (Ref. 1: Osiko, V. V., Maksimova, G. V., Optika i spektroskopiya, 9, vyp. 4). It is noted here that the relationship between the luminescence properties and the mean valence of Mn permits the lumincphores under investigation to be classified into three groups. The valence of Sn has been studied on a group of phosphate phosphors. Results are presented in Card 1/4

22155

Valence of Sn and M

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Table 1. Tin is shown to be in the bivalent state in all luminophores, regardless of the differences in the luminescence spectra. An oxidation of tin, causing it to pass over into the tetravalent state, and also its reduction to the elementary state, cause luminescence to disappear. It has been further established that a reducing atmosphere is not in all cases necessary to produce a bivalent tin: some luminophores are produced also by sintering in the gir.. Their composition includes, however, a reducing agent (ammonium salts, which are decomposed at t = 1000°C with hydrogen being liberated). In the study of the relationship between valence of the activator in the luminophore and the temperature, on the one hand, and sintering in our and concentration of the activator, on the other, the authors examined the system ZnO-MnO-02 which was regarded as a physicochemical model of a luminophore. The specimens produced from mixtures of ZnO and MnO2, were sintered at different temperatures in an oxygen flow or in a nitrogen flow purified from oxygen. The mean valence of Mn was determined on the resulting specimens. Results are presented in Fig. 1. It shows that the mean valence of Mn is strongly dependent, under equal conditions, on the relative Mn content. It may be also seen that up to a given concentration, the mean valence is 2 and independent of the sintering

Card 2/4

Valence of Sn and Mn atmosphere. This confir determined by the crysta and the same luminophore production, a different must be taken into account activator is connected where also able to show the with high Mn content in cition: Besides the (Zn, Balso manganese oxide is a discounted to the content of the content of the content of the content oxide is a discountered.	may have, un Mn valence, d nt that a cha ith changes i hat when sint oxygen, these	is case the structure. It der otherwise epending on tage of the van the phase cering the sol single-phase	t also follows equal condition on tending the Mn contended state composition. It is systems under the content of the content o	is the tions t. The (Zn, ergo	t one of he fact s authors Mn)0 lamina-	
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Legend to Table 1: Juminescence color; Sintering conditions:	Inmunopop (Sr. Mg) ₂ (PO ₄) ₄ —	the microsco	2) CAUSER S SPOKERS	re 1	figure, Basess.3)	20
egend to Table 1:) luminescence color;) sintering conditions;	JEMES OF OF	VIE MICTOSCO	2) словия прокади.	E'/,	figure, Basess.3)	20

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5/048/61/025/004/005/048 B104/B201

AUTHORS:

Corbacheva. N. A., Gugel', B. M., Konstantinova-Shlezinger, M. A., Lapir, Ye. S., and Rutshteyn, T. G.

TITLE:

Phosphate luminophores for luminescent lamps with improved

light emission

PERIODICAL: Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, v. 25,

no. 4, 1961, 455-458

The present paper has been read at the 9th Conference on Luminescence (Crystal Phosphors), Kiyev, June 20-25, 1960. In addition to the requirement that luminophores should have a "white" spectrum, also that of the quantum yield to be as high as possible should be satisfied. Barium-titanium-phosphate (BTP) and strontium-magnesium-phosphate (SMP), which satisfy these requirements best, are the object of the present study. The temperature stability of SMP was improved by the introduction of B203 to such an extent as to make it suitable for correcting the color of high pressure Hg lamps. BTP was prepared by a three-hour sintering of a

Card 1/6

22156

Phosphate luminophores for ...

S/048/61/025/004/005/048 B104/B201

mixture of BaHPO₄, TiO₂, and BaF₂ at 1075°C. Data regarding the brightness and the stability of luminescence of the specimens concerned are presented in Tables 1 and 2. As may be seen from Table 1, Mn acts as an extinguisher. The spectral composition of emission is shown in Fig. 1. As may be noted from the tables, a BaO excess reduces brightness strongly, whereas a P₂O₅ excess (up to 5 moles) has no effect whatever. A BaO excess leads to the formation of 4BaO:3TiO₂°P₂O₂, whereby the activator concentration is reduced. It is found, furthermore, that the introduction of BaO certain concentration is attained. SMP was prepared by three different sintering methods from mixtures SrCO₃, MgCO₃, (NH₄)₂HPO₄, and SnO₂.

- 1) One-hour sintering at 600°C in air, and, after grinding, renewed two-hour sintering at 1200°C, and, finally, at 1200°C for 30 minutes in NH3.
- 2) Heating from room temperature to 1200°C in one hour, and a second sintering at 1200°C for 30 minutes in NHz or with addition of carbon.

 3) Heating of phosphates and carbonates (without 5)

3) Heating of phosphates and carbonates (without Sn) from 20 to 1200°C in one hour, crushing together with H₃C·SnO·OH and sintering in a closed tube

Card 2/6

10:

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Phosphate luminophores for... \$\frac{22156}{5/048/61/025/004/005/048}
B104/B201

at 1200°C for 30 minutes. The luminophore had the composition (Sr_{2.63}Mg_{0.34})(PO₄)₂Sn_{0.04}. Spectra of different luminophores are graphically presented in Fig. 1. Data regarding the effect of the production method upon the luminophore quality are given in Table 2. Data of temperature stability are graphically presented in Fig. 2. 15 w luminescent lamps with improved light emission and a light temperature of 4500°K, possessing a Harrison factor of 86 % and a light yield of 34 lm w-1, were prepared from a mixture of 50 % BTP and 50 % SMP. If a mixture of 70 % calcium halogen phosphate (activated with Sb) and 30 % SMP is used, a lamp with a light temperature of 6500°C, with equal Harrison factor, and equal light yield can be obtained as is the case in industrial luminescent lamps the luminophore of which is made of a mixture of 85 % calcium halogen phosphate (activated with Sb and Mn) and 15 % magnesium arsenate (activated with Mn). There are 2 figures, 2 tables, and 4 references: 1 Soviet-bloc and 3 non-Soviet-bloc.

Card 3/6

GCREACHEVA, N.A.; KONSTANTINOVA-SHLEZINGER, M.A.

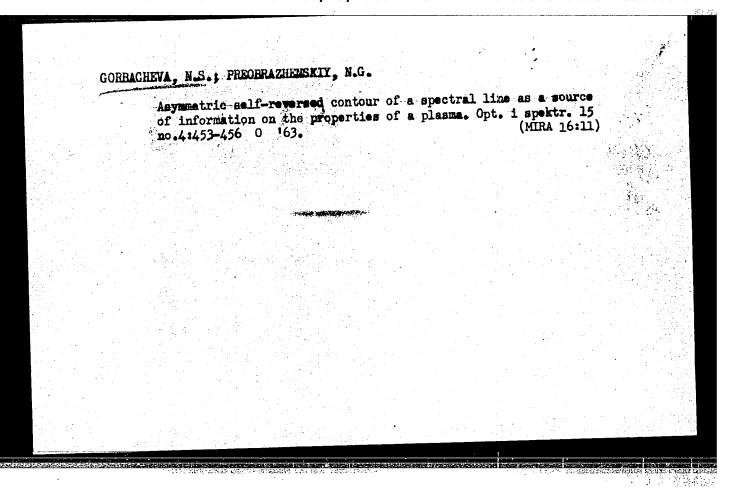
Determining uranium by measuring the width of the luminescent zone in a chromatogram. Zhur. prikl. spekt. 3 no. 2:172-174. Ag '65. (MIRA 18:12)

1. Submitted Oct. 13, 1964.

ACC NR: AP7004991	SOURCE CODE: UR/004	18/66/030/009/1521/1523	
AUTHOR: Gorbacheva, N.A.			
ORG: none		A second according	
TITLE: Europium-activated and magnesium orthophosphate lumino Luminescence (Crystal Phosphors	held at Riga, 16-23 Sep	t. 196 <u>5.</u> 7	
Source: An SSSR, Izvestiya. Se	riya fizicheskaya, v. 30,	no.9, 1966, 1521-1523	
TOPIC TAGS: luminescence, strochemical reduction	ntium, msgnesium phosphor	us, europium, manganes	
ABSTRACT: The author investig the nature of the yellow lumine materials and is not usually as heating a mixture containing (Nates in a molar ratio of 89.11, (PO ₄) ₃ , and Sr,Ng ₂ (PO ₄) ₂ were on the synthesis conditions. materials were recorded. The taken place in a rapid stream NH ₃ stream there was observed	cribed to Eu2+. The material of the stream of the control of the stream of the control of the co	erials were synthesized trontium and magnesium of NH3. Sr3(PO ₄)2, Mg3 in varying amounts, dep ssion spectra of all the mitted when the synthesis had taken place in a	carbon- bending bese sis had slow
Card 1/2			
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AP7004991 ACC NR. taneous presence of di- and trivalent europium. X-ray studies showed that the yellow and violet luminescing materials belonged to the same high-temperature modification of strontium orthophosphate. It is therefore concluded that the yellow luminescence is due to some auxiliary microchanges. Two hypotheses are advanced to account for the yellow luminescence: 1) the europium is reduced in a strong NH3 stream to a nonvalent state; and 2) microdefects arise in the neighborhood of the Eu2+ centers and form new centers with these ions. Arguments for and against both these hypotheses are presented. The author favors the second hypothesis, but she points out that the two hypotheses are not mutually exclusive. Phosphate luminophors simultaneously activated with Eu and Mn are discussed briefly. By heating these materials in a reducing atmosphere one can obtain phosphors that luminesce in the red. Mg3(PO4)21 Eu: Mn and SrMg_(PO4)2: Eu: Mn are of technical interest because they can be strongly excited at 2537 Å. These phosphors are very temperature sensitive, however; when These phosphors are very temperature sensitive, however; when the reason for this has been learned, the development of the materials will be continued. The author thanks M.A. Konstantinova-Shlezinger for discussions and L.M. Tsyganov for assistance with the work. 001

Construction of gas lines in Krasnodarsk Territory. 6 no. 2:21-22 F 161. (Krasnodarsk Territory—Gas pipes)	Stroi. truboprov. (MIRA 14:5)



S/081/63/000/004/033/051 B194/B180

海斯 新黎

AUTHORS: Grodzovskaya, R. I., Gorbacheva, N. V.

TITLE: Sulfonation of extracts from the selective refinement of

lubricating oils

PERIODICAL, Referativnyy zhurnal. Khimiya, no. 4, 1963, 519, abstract
4P134 (Novosti neft. i gaz. tekhn. Neftepererabotka i nefteu

khimiya", no. 8, 1962, 8-10)

TEXT: To obtain the de-emulsifier HYK (NChK), solutions of diesel fuel fractions containing 10 and 30% of extracts from the purification of distillate and residual oils were sulfonated (direct sulfonation of extracts is not possible because of the high density of the resulting tar, while higher concentrations of extract in the solution cause the formation of acid sludges the consistency of which makes transportation impossible). Sulfonation was carried out with 45 and 50% H₂SO₄ at 70 - 75° in two stages,

two hours in each. It was shown that the yield of NChK obtained by sulfunating a starting material containing dissolved extracts rises with the concentration of the extract. The H₂SO₄ consumption per ton NChK falls

Card 1/2

Sulfonation of extracts from...

S/081/63/000/004/033/051-B194/B180

correspondingly. An increase of up to 50% in the quantity of H₂SO₄ used for the starting material does not increase its specific consumption on NChK because the sulfonation is deeper. A check on laboratory results under commercial conditions has shown the NChK yield to be 50% higher and the H₂SO₄ consumption per ton of de-emulsifier more than 30% less, than when projuced from dieselfuel. The salting-out in NChK obtained by sulfonating diesel fuel containing 30% extracts from distillate oil refinement (2009 (ELOU)) was no worse than in NChK obtained from the conventional starting material. (4b-stracter's note: Complete translation.)

Card 2/2

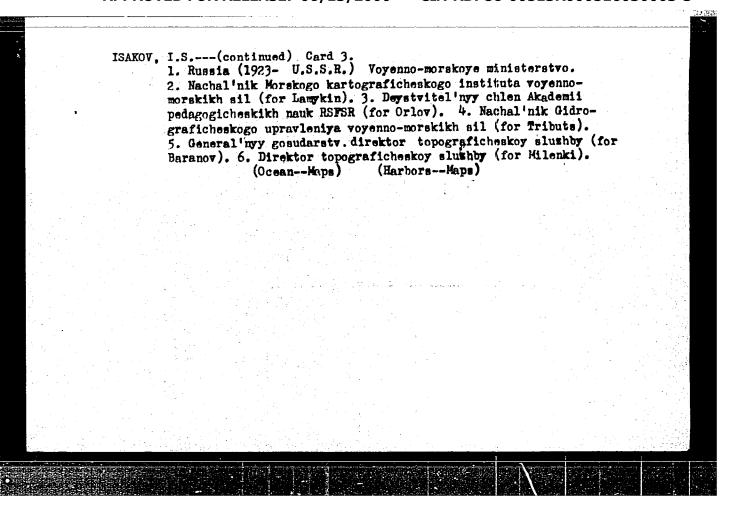
GORBACHEVA, N. YE.

ISAKOV, I.S., prof., admiral flota, otv.red.; PETROVSKIY, V.A., dotsent. kand.voyenno-morskikh nauk, kontr-admiral, red. [deceased]; DEMIN, L.A., dotsent, kand.geograf.nauk, insh.-kapitan 1 ranga, glavnyy red.; BARANOV, A.N., red.; BERG, L.S., akademik, inzh.-mayor, red.; BOLOGOV, N.A., dotsent, kontr-admiral v otstavke, red.; VITVER, I.A., professor, doktor geograf.nauk, red.; GRIGOR'YEV, A.A., akademik; YEGOR'YEV, V.Ye., zasluzhennyy deyatel' nauki, prof., doktor voyenno-morskikh nauk, kontr-admiral v otstavke, red.; ZIMAN, L.Ya., prof., red.; ZUBOV, N.N., prof., doktor geograf. nauk, inzh.-kontr-admiral v otstavke, red.; KAVRAYSKIY, V.V., prof., doktor fisiko-mat.nauk, inzh.-kontr-admiral v otstavke, red.; KALESNIK, S.V., prof., doktor geograf.nauk, red.; KUDRYAVTSEV, M.K., general-leytenant tekhn.voysk, red.; LAMYKIN, S.M., kapitan 1 ranga, red.; MATUSEVICH, N.N., saslushennyy deyatel' nauki i tekhniki, prof., doktor fiziko-mat.nauk, insh.-vitse-admiral v otstavke, red.; [deceased]: MESHCHANINOV, I.I., akademik, red.; MILENKI, S.G., red.; ORLOV, B.P., prof., doktor geograf.nauk, red.; PANTELEYEV, Yu.A., vitse-admiral, red.; SNEZHINSKIY, V.A., dotsent, kand.voyennomorskikh nauk, insh.-kapitan l ranga, red.; SALISHCHEV, K.A., prof., doktor tekhn.nauk, red.; TRIBUTS, V.F., admiral, red.; FOKIN, V.A., vitae-admiral, red.; SHVEDE, Ye.Ye., prof., doktor voyenno-morskikh nauk, kontr-admiral, red.; SHULETKIN, V.V., akademik, insh.-kapitan 1 ranga, red.; PAVIOV, V.V., inzh.-polkovnik, red.; VOLKOV, F.G., (Continued on next card)

ISAKOV, I.S.,--(continued) Card 2.

podpolkovnik, pomoshchnik glavnogo red. po izd-vu; SEDOV, N.Ye.,
kapitan 2 ranga, uchenyy sekretar'; VOROB'YEV, V.I., kapitan
l ranga, red.kart; MIGALKIN, G.A., inzh.-kapitan l ranga, red.kart;
GAPONOVA, A.A., red.kart; GONCHAROVA, A.I., red.kart; GORBACHEVA,
N.Ye., red.kart; GHYUNBERG, G.Yu., red.kart; DUROV, A.G., red.
kart; TERSHOV, I.B., red.kart; ZIL'BERSHER, A.B., red.kart;
KASTAL'SKAYA, N.I., red.kart; KUBLIKOVA, M.M., red.kart; MAKAROVA,
V.N., red.kart; MOROZOVA, A.F., red.kart; PAVLOVA, Ye.A., red.
kart; POCHUBUT, A.N., red.kart; ROMANOVA, G.N., red.kart; SMIRNOVA,
L.V., red.kart; SMIRNOVA, L.N., red.kart; TANANKOVA, A.I., red.
kart; YANEVICH, M.A., red.kart; YASINSKAYA, L.F., red.kart;
VASIL'YEVA, Z.P., tekhn.red.; VIZIROVA, G.N., tekhn.red.; GOLOVANOVA,
A.T., tekhn.red.; GOROKHOV, V.I., tekhn.red.; MALINKO, V.I., tekhn.
red.; SVIDERSKAYA, G.V., tekhn.red.; CHERNOGOROVA, L.P., tekhn.red.;
FURAYEVA, Ye.M., tekhn.red.;

[Marine atlas] Morskoi atlas. Otv.red. I.S. Isakov. Glav.red.
L.A. Demin. Isd. Morskogo general'nogo shtaba. Vol.1 [Navigation geography] Navigatsionno-geograficheskii. Zamestitel' otv. red.
po I tomu V.A. Petrovskii. 1950. 83 maps. (MIRA 12:1)
(Continued on next card)



GOL'DENBLAT, Iosif Israilevich; GORBACHEVA, O.S., redaktor; MURASHOVA,

H.Ta., tekhnicheskiy redaktor.

[Problems of the mechanics of deforming media] Mekotorye voprosy
mekhaniki deformirusnykh sred. Moskva, Gos.isd-vo tekhnikoteoret. 19t-ry. 1955. 271 p.

(Deformations(Mechanics))

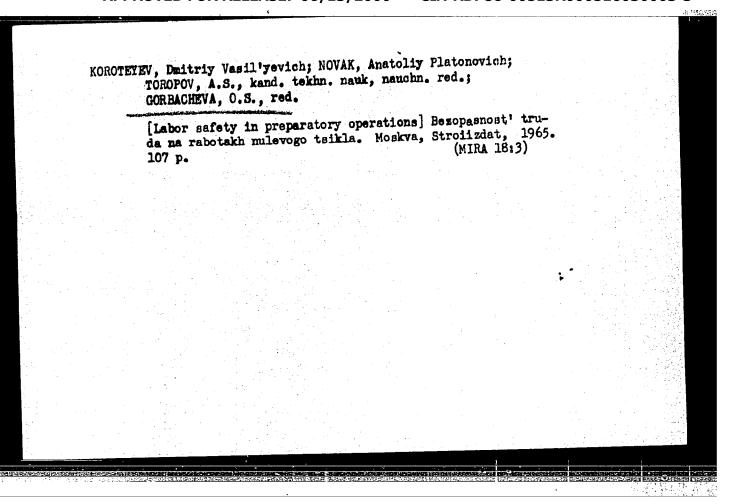
(MLRA 8:12)

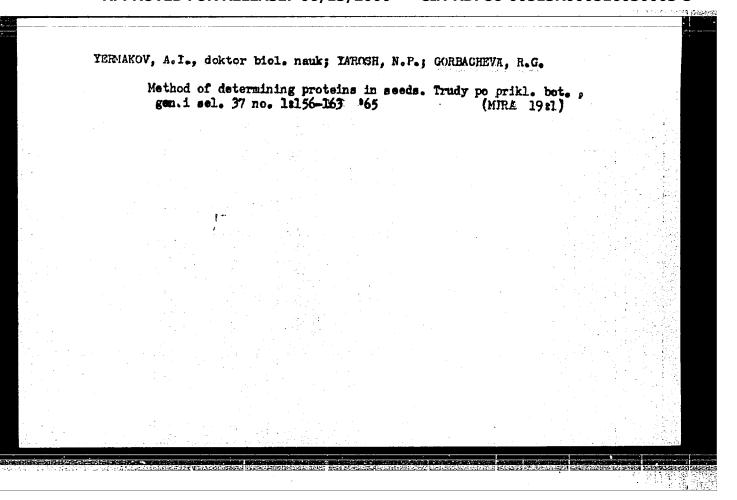
LUR'YE, Anatoliy Isaakovich; QORBACHEVA, O.S., redaktor; 'ZUMARKINA, N.A., tekhnicheskiy redaktor.

[Spatial problems in the theory of elasticity] Prostranstvennye sadachi teorii uprugosti. Moskva, Gos.izd-vo tekhniko-teoret.

1it-ry, 1955. 491 p.

(Elasticity)





Wordacheur, S. W.	
ALEKSANDROVA, F.A.; GORBACHEVA, S.G.	
Increasing the output of lubricant-producing installat improving laboratory control. Proisv. smas. mat. no.3:	ions by 8-9 '57. (MIRA 10:12)
1. Pervyy Moskovskiy neftemslosavod. (Moscow-Imbrication and lubricants)	

YEFIMOV, A.F.; GORBACHEVA, T.B.

Potassium feldspars in the alkali pegmatites of the Inaglinskiy massif. Trudy Min. mus. no.14:231-237 '63. (MIRA 16:10)

(Inaglinskiy massif—Feldspar)
(Inaglinskiy massif—Fegmatites)

1. 24674-65 EWP(a)/EPA(s)-2/EWT(m)/EPF(c)/FCS/EWG(v)/EPR/FWP(j)/T/EWP(b)/EWA(l)
PC-4/Pa-5/P1-4/Pr-4/Ps-4/Pt-10 RM/WW

ACCESSION NR: AP5004667

5/0191/64/000/009/0013/0017

AUTHOR: Severov, A. A., Gorbacheva, T. B.; Lukin, B. V.; Sergeyev, V. K.

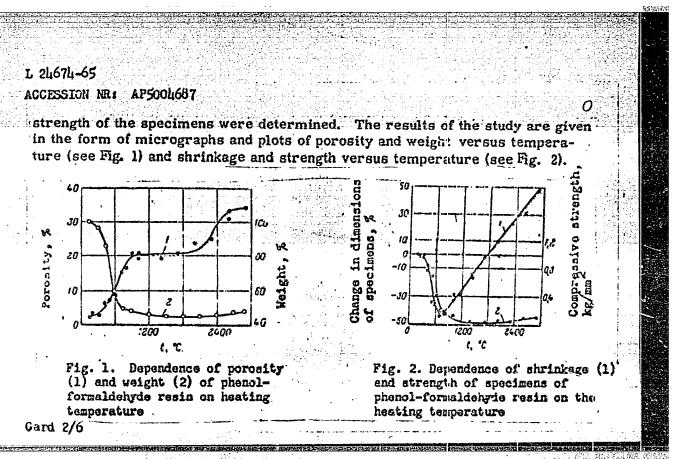
TITLE: Changes in the fine and porous structures of phenol-formaldehyde resin during rapid short-duration heating to high temperatures

SOURCE: Plasticheskiye massy, no. 9, 1964, 13-17

TOPIC TAGS: phenolic plastic, polymerization, heat effect, crystal chemistry, polymer structure

Abstract. Changes in the structure of GOST 4559-49 phenol-formaldehyde resin. have been studied during rapid short-duration heating up to 290° C. The initial resin was cured for about 20 days at 160° C. its degree of polymerizations as 36.2%. The specimens were heated at rates of 10.000-20.00° C/min. ideating was conducted in increments of 100° below 1100° C and 300° above 100° C, with a 1-min holding time at each temperature. The samples were then cooled in nitrogen. Changes in the porous structure of the specimens were studied by visual observation, micrographs, and porosity measurements based on moisture absorption. In addition, weight loss, shrinkage, and compressive

Card 1/6

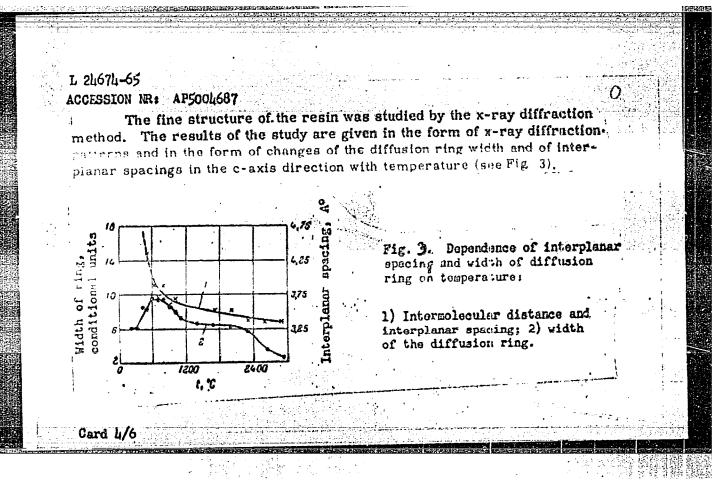


L 24674-65

ACCESSION NR: AF5004687

The results showed that: 1) Pores and cracks develop rapidly at 400-700° C as a result of the evolution of volatile pyrolysis products. process causes considerable weight loss and shrinkage of specimens. 2) The pores continue to develop at 700-1300°C, but at a slower rate. At the same time wide cracks are formed. These cracks cannot be determined by moisture absorption, and the magnitude of the measured porosity remains unchanged up to 1900° C. 3) At 1900-2600° C, the pores continue to develop; since specimen weight remains unchanged, it is concluded that the porosity develops as a result of an increase in the density of the coke pore walls. 4) At 2600-2900° C. the pores become filled with secondary products formed by pyrolysisproduct decomposition. The specimens become blocks and acquire a metallic luster, and their weight increases slightly. 5) The specimen volume increases continuously at above 700° C and attains 150% of its initial value at 2900° C. 6) The specimen compressive strength drops from its initial value of $700-2100 \text{ kg/mm}^2$ to 0.05 kg/mm^2 at $1700-2600^\circ$ C. and then increases again at 2900° C to 0, 10 kg/mm² owing to the deposition of secondary products which fill the pores and cracks.

Card 3/6



L 24674-65

ACCESSION NR: AP500L687

These results show that: 1) Heating of the resin to 250° C causes its further polymerization. 2) At 300-700° C, the resin degrades and coke structures are formed. 3) Above 800° C, the formation of primary and the ordering of secondary coke structures (bundles) continues; the two-dimensional coke-structure formation ends at 1200-1300° C. 4) At 1200-2300° C, slow growth of bundles continues. 5) At higher temperatures, in the pregraphitization period, the bundles begin to grow more rapidly; regions with a three-dimensional ordering (crystallites of graphite) appear at 2900° C. Thus during rapid heating graphitization begins at higher temperatures than during heating at a rate of 10° C/min with 2-hr holding periods, in which case graphitization begins / at 2400° C.

COMMENT: The article is interesting as an apparent attempt to determine the character and possibly the rate of progressive thermal deterioration of a GRP binder at temperatures and heating rates comparable to those arising in missile combustion chambers or on the surface of re-entry plates. At the given heating rate, i.e., 170-330 C/sec, testing temperatures of 100-2910°C could be reached within the time required to reproduce approximately the thermal conditions to which GRP used for serospace purposes is subjected. It is true that only the binder and not the GRP itself was tested, and that heat transfer was not

Card 5/6

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The low compressi	ve-strength values lusting the crumbl	ing of coked			
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GORBACHEVA. T.B.; YEMEL'YANENKO, P.F.

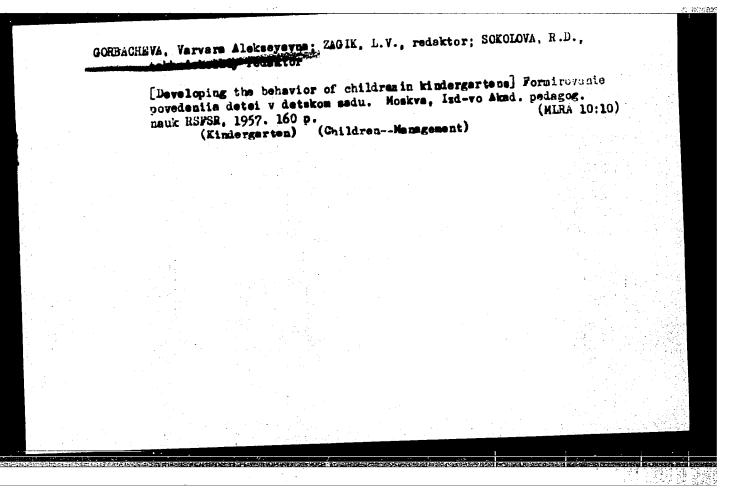
Potash feldspar in the Inagli intrusive (Aldan Plateau). Vest. Mosk.un. Ser. 4: Geol. 19 no. 5:47-54 S-0 164. (MIRA 17:12)

1. Kafedra dinamicheskoy geologii Moskovskogo universiteta.

SHLIONSKIY, Sh.G.; GORBACHEVA, V.A.

Computation of the lowest applicable frequencies and other short-wave communication quantitatives by means of electronic computers. Geomag. i aer. 3 no.4: 711-716 Jl-Ag '63. (MIRA 16:11)

l. Institut zemnogo magnetizma, ionosfery i rasprostraneniya radiovoln AN SSSR.



Extermination of weeds by chemical substances. Izv. AN Turk. SSR. Ser. biol. nauk no.4:77-80 '63. (MIRA 16:9)

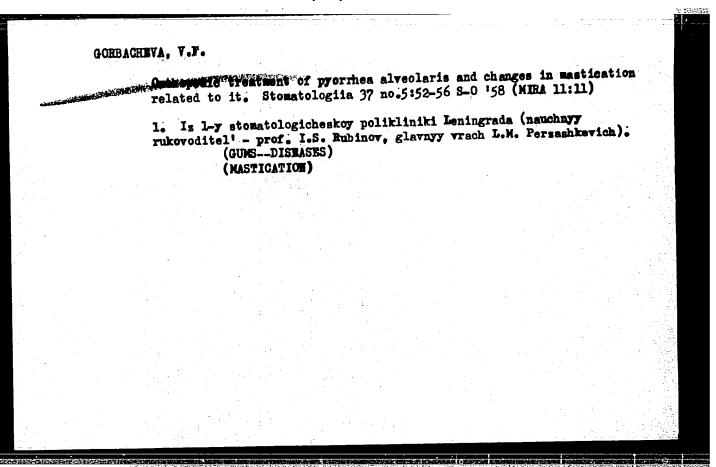
1. Ashkhabadskoys otdeleniye Sredneaziatskoy zheleznoy dorogi.
(Ashkhabad region-Weed control) (Herbicides)

GORBACHEVA, V. F.: "The physiological investigation of the functions of cheming and certain aspects of the orthopedic treatment of amphedontosis." Acad Sci USSR. Inst of Physiology imeni I.P. Pavlov. Leningrad, 1956. (Dissertation for the Degree of Candidate in Medical Sciences).

Source: Knizhnaya letopis! No. 28 1956 Moscow

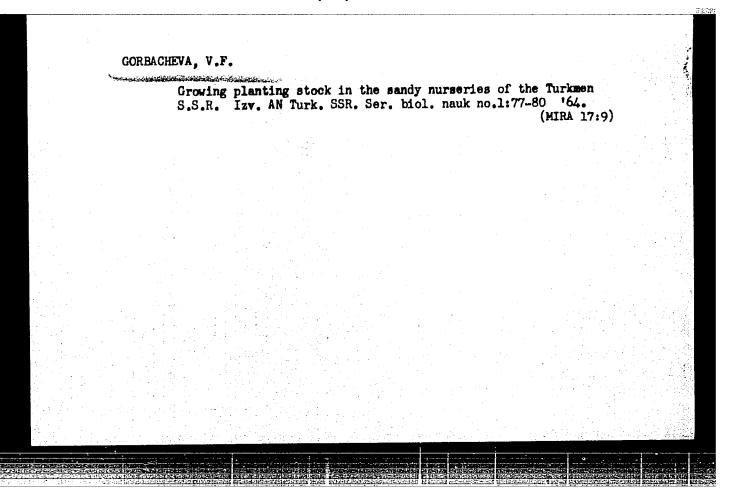
Bole of orthodontic measures in the treatment of pyorrhes alveolaris.
Stomatologile 36 no.2:59-61 Mr-Ap '57. (MIRA 10:6)

1. Is Leningredskoy gorodskoy stomatologicheskoy polikliniki (Nauchnyy rukovoditel' - prof. I.S.Rubinov, glavnyy vrach L.M.
Persashkevich)
(GUNS-+DISEASES) (TEETH--ABNOIMITIES AND DEFORMITIES)

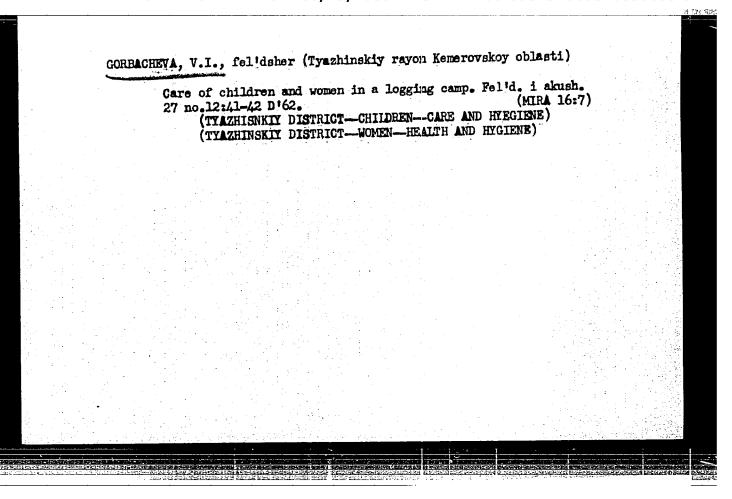


17年1月

Dynamics treatment 160.	of the change in of parodontosis	mastication in t (amphodontosis).	he process o Trudy LSGM	f orthopedic I 63:71-80 (MIRA 15:1)	
(G	UMSDISEASES)	(MASTICATION)			
. 4.					



Binding sandy soils with polyacrylamide emulsion. Izv. AN Turk. SSSR. Ser. biol. nauk no.2:57-60 '64" (MIRA 17:6)
1. Ashkhabadskoye otdeleniye Sredneaziatskoy zheleznoy dorogi.



GORBACHEVA, V.O.

AUTHOR:

Gorbacheva, V.O., and Mikhaylov, N.V.

69-20-1-6/20

TITLE:

Structure and Phase State of Polyethylene Terephthalate Fibers (O strukture i fazovom sostoyanii volokon iz polietilentereftalata)

PERIODICAL: Kolloidnyy Zhurnal, 1958, Vol. XX, # 1, pp 38-42 (USSR)

ABSTRACT:

Polyethylene terephthalate is a new polymer, which has been widely used in industry lately. It is employed in the production of synthetic fibers, films and plastics. It is resistant to acids, oxygen, light, bacteria, has a low permeability for gases and a sufficient mechanical resistance at various temperatures. In the article the structure of polyethylene terephthalate and fibers made from it by various processes are investigated. The specimens under investigation have a viscosity of 0.24-0.27. The fiber made from it was spun at 285°C and was stretched at 80-100°C to 550%. X-ray and thermographic analyses were made. Fig. 1.1, shows the roentgenogram of polyethylene terephthalate in the form of a solid transparent mass. This roentgenogram shows the unstretched fiber. The roentgen picture of both specimens is the same and is characterized by a broad interference, which is an indication of

Card 1/3

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-005123R000546030003-3

Structure and Phase State of Polyethylene Terephthalate Fibers

the amorphous structure of the substance. The phase state alone is no sufficient criteria for determining the structure of a polymer. The thermographic method of phase analysis was therefore also used. In fig. 2 a, the differential curves of heating and cooling of the initial polyethylene terephthalate. The thermographic picture changes, if the initial polymer and the unstretched fiber are preliminarily heated and the fiber stretched at increased temperature. In fig. 2 b, the thermograms of such specimens are presented. During heating of the polymer devitrification takes place and the kinetic energy of the links is increased. During devitrification or after it, in the solid state of the substance, crystallization takes place. In the heating curve, therefore, an exothermic effect in the temperature interval 105-150°C is observed (Fig. 2 a, area B). The obtained thermographic data was used for determining the melting heat of polyethylene terephthalate, which is 9-11 kcal/g. The crystallization of the polymer takes place at temperatures of 80-110°C.

Card 2/3

MIKHAYLOV, H.V.; BUKOV, G.A.; COMPACHIVA. V.O.; MAKAROVA, T.P.; v rabote prinimali uchastiye: IARIOHOV, P.E.; SOROKHA, V.I.; ZOTOV, Ya.E.

Studying the formation mechanism of synthetic fibers from molten materials. Unim.volok. no.1:33-36 '59. (MIRA 12:8)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo volokna. (Textile fibers, Synthetic)

MIKHAYLOV, N.V.; FAYMBERG, E.Z.; GORRACHEVA, V.O.

Study of the molecular structure of stereoregular polymers.
Isotatic polypropylone. Vysokom.soed. 1 no.lf143-148 Ja '59.

(MIRA 12:9)

1. Vsesoyusnyy nauchno-issledovatel'skiy institut iskusstvennogo velena.

(Polymers) (Prepene)

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000516030003-3"

MIKHAYLOV, N.V.; SHEYN, T.I.; GORBACHEVA, V.O.; TOPCHIRASHEVA, V.N.; vrabote prinimali uchastive temperature laboranty; IARIOROV, P.M.; VLASOVA, L.P.; MURASHKINA, S.I.

Investigating the molecular structure of synthetic fibers. Part 14: Physicochemical and physicomechanical properties of the polycapramide - polyundecanamide polyamide group. Vysokom. soed. 1 no.2:185-190 F 159. (MIRA 12:10)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskuestvennogo volokna.

(Textile fibers, Synthetic) (Amides)

MIKHAYLOV, N.V.; ZAV'YALOVA, N.N.; GORBACHEVA, V.O.

Oradient method of determining the specific gravity of synthetic fibers. Khim.volok. mo.1:19-22 '60. (MIRA 13:6)

1. Vsesoyusnyy nauchno-issledovatel'skiy institut iskusstvennogo volokma.

(Textile fibers, Synthetic) (Specific gravity)

S/183/60/000/005/007/007 B028/B054

AUTHORS:

Demina, N. V., Gorbacheva, V. O., Kotina, V. Ye.,

Ukhanova, Z. V.

TITLE:

Properties of Chemical Fibers

PERIODICAL:

Khimicheskiye volokna, 1960, No. 5, pp. 40-41

TEXT: This paper describes testing methods for chemical fibers. All mechanical properties of fibers were tested at an air moisture of 65±1% and an air temperature of 20±2°. The following testing methods are indicated: 1) Control of stability and elongation of threads on pendulum-type tensile-testing machines at a distance of 500 mm between the strainers and an average time until breaking of 15 sec. FOCT 6611-55 (GOST 6611-55). Impact tensile-testing machines were used for staple fibers. 2) The deformation (expansion) modulus was determined from the ratio between load and relative deformation for threads elongated by %.

3) Elasticity of threads was tested by a dynamometer. Threads were stretched by 4% and 10% of their original length, left in this state for one minute, and relaxed for one minute; the remaining elongation was measured.

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Properties of Chemical Fibers

S/183/60/000/005/007/007 B028/B054

4) The shearing modulus was checked by torsional vibrations with a KM-20 (KM-20) pendulum-disk device. 5) Bending-stress durability was tested with a ATT-15 (DP-15) device at 110 cycles per minute and a stress of 5 kg/mm2. A "Sinus" device was used for elementary fibers at a stress of 10 kg/mm2. 6) Wear resistance was tested by grinding a thread until breaking on a corundum disk at 160 rpm. 7) Stability to ultraviolet light was determined by 20 hours' irradiation with a TPK-2 (PRK-2) mercury vapor lamp. 8) The elasticity of the fiber mass was tested in a cylinder by . volume change under a load of 70 kg for 60 min. Relaxation lasted 30 min; the remaining volume was measured. 9) Moisture content of the fiber by absolute drying in a drying chamber at 105-110°C (chlorine fibers at 70°C). The material had been previously stored for some time at an air moisture of 65% and a temperature of 20+2°C. Data are given in % referred to the total dry substance. 10) The specific heat was determined by an adiabatic calorimeter. 11) A differential thermal analysis yielded data on the temperature range of melts and crystallization. A table comprising 14 pages lists results and X-ray patterns of fibers of viscose, acetate cellulose, caprone, anide, enanth, pelargone, undecane, polyamides,

Card 2/3

Properties of Chemical Fibers

S/183/60/000/005/007/007. B028/B054

polyesters, polyolefins, chlorine, polyphene, nitron, acrylonitrile, acetochlorine, and vinitrone. The laboratory assistants A. V. Poluyanova, T. I. Negina, and Ye. P. D'yakova cooperated in the investigations. There are 1 table and 6 Soviet references.

ASSOCIATION: VNIIV (All-Union Scientific Research Institute of Synthetic Fibers)

Card: 3/3

DEMINA, N.V.; GORBACHEVA, V.O.; KOTINA, V.Ye.; UKHANOVA, Z.V.

Properties of synthetic fibers. Khim.volok. no.5:40-55 '60.
(MIRA 13:12)

1. Vsesoyusnyy nauchno-issledovatel'skiy institut iskusstvennogo volokna.

(Textile fibers, Synthetic)

•*	Structure of polyamides obtained by interfacial polycondensation. Tysokom. soed. 2 no.8:1283-1286 Ag'60. (MIRA 13:9)
	l. Vsesoyusnyy nauchno-issledovatel skiy institut iskusstvennogo volokna.
	(Polyamides)

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000516030003-3"

5/190/62/004/002/011/021 Mikhaylov, N. V., Faynberg, E. Z., Gorbacheva, V. O. Chieng Compatibility of the system polyethylene - polypropylene Vysokomolekulyarnyye soyedineniya, v. 4, no. 2, 1962, Ch'ing-hai AUTHORS: TEXT: A method of combining polyhydrocarbons from their solutions has been developed. A mixture of low-density polyethylene (PE) and isotact TEAT: A method of combining polynydrocarbons from their solutions has been developed. A mixture of low-density polyethylene or white spirit or melf with polypropylene (PP) was produced via o-xylene or white spirit. been developed. A mixture of low-density polyethylene (PE) and isotactic with polypropylene (PP) was produced via 0-xylene or white spirit or melt with 160 - 1650 polypropylene (PP) was produced via 0-xylene or so min at t = 160 - 1650 polypropylene (PP) ratios. Dissolution took AO - 50 min at t = 160 - 1650 polypropylene (PP) ratios. polypropylene (PP) was produced via o-xylene or white spirit or melt with 165°C different PE; PP ratios. Dissolution took 40 - 50 min at t = 160 cooling (total concentration = 0.1; 0.5; 5%). The precipitate formed by cooling (total concentration = 0.1; 0.5; 5%). TITLE: different PE; pp ratios. Dissolution took 40 - 50 min at t = 160 - 165°C min at t = 160 - 160°C min at t = 160°C mi PERIODICAL: (total concentration = 0.1; 0.5; 5%). The precipitate formed by cooling The physicoThe physical remove the solvent. The physical to 80 - 85°C was eluted with acetone to remove studied by (a) differential to 80 - 85°C was eluted with acetone were studied by (b) differential to 80 - 85°C was eluted with acetone were studied by (a) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (a) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (a) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (b) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied by (c) differential to 80°C was eluted with acetone were studied with acetone were to bu - by twee eluted with acetone to remove the solvent. The physico-themical properties of polymer mixtures (c) density measurement. The physico-thermal analysis: chemical properties of polymer mixtures were studied by (a) differential The (c) density measurement. The (c) density measurement and copolymers and copolymers and copolymers thermal analysis; (b) thermal curves for pure polymers and correspond to the heating curves of melting. The two endothermic effects of the heating curves of melting. endothermic effects of the heating curves for pure polymers and copolymers.

The two endothermic effects of melting. The two endothermic of melting. The temperature range of the temperature range of the curves for polymer mixtures correspond to the temperature. Card 1/3

Compatibility of the system ...

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the transition of pure polymers, and only for mixtures 7.5: 2.5; 8:2; 9: 1; 9.5: 0.5; and 9.8: 0.2, they showed only one endothermic effect, like the curves for the initial polymers. The concentration range of compatibility is limited; concentration decrease of PE and increase of PP effect demixing. Since the temperature range of melting of copolymers only differs by 15 - 18°C from that of pure PE, the difference should be even smaller for combined mixtures. This also agrees with Flory's idea on the decrease of the melting point when plasticizing one polymer by another (lowor high-molecular). Comparisons of the heat capacity with the values of the copolymer are used as a criterion for the degree of combination of polymer mixtures. The heat capacities of pure homopolymers are close to each other, and strongly differ from those of copolymers. The polymer mixture 8: 2 has maximum heat capacity and optimum compatibility. Minimum density (0.915) of the copolymer corresponds to maximum heat capacity (0.500). The copolymer has a lower than the additive density, and thus a molecular packing of lower density. The density of all combined mixtures is lower than the additive value. The mixture 7.5: 2.5 shows maximum deviation. This proves a plasticizing effect of PE on PP owing to higher flexibility of the polymer chains of PE. This effects a decrease in stiffness of PP, and facilitates its compatibility with PE. There are Card 2/3

CIA-RDP86-00513R000516030003-3 "APPROVED FOR RELEASE: 06/13/2000

Compatibility of the system ...

5/190/62/004/002/011/021 B110/B101

3 figures, 2 tables, and 3 references: 2 Soviet and 1 non-Soviet. The reference to the English-language publication reads as follows: R. Kee, J. Polymer Sci., 42, 15, 1960.

ASSOCIATION: Nauchno-issledovatel'skiy institut iskusstvennogo volokna

(Scientific Research Institute of Synthetic Fibers)

SUBMITTED:

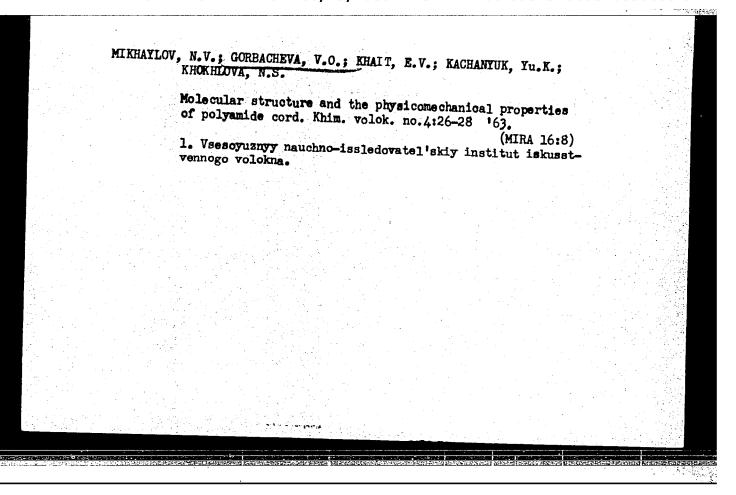
February 9, 1961

Card 3/3

AFANAS'YEVA, G.N.; VOL'F. L.A.; MEOS, A.I.; GORBACHEVA, V.O.; MIKHAYLOV, N.V.; MIL'KOVA, L.P.

Thermoplasticization stretching of polyvinyl alcohol fibers.
Khim. volok. no.5:16-19 '63. (MIRA 16:10)

1. Leningradskiy tekstil'nyy institut imeni S.M. Kirova (for Afanas'yeva, Vol'f, Meos). 2. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo volokna (for Gorbacheva, Mikhaylov, (Mil'kova).



MIKHAYLOV, N.V.; GORBACHEVA, V.O.; IYEVLEVA, A.K.

Determination of the specific volumes of synthetic fibers at elevated temperatures. Khim. volok. no.5:26-28 '63.

(MIRA 16:10)

1. Vsesoyuznyy nauchn-issledovatel'skiy institut iskusstvennogo volokna.

GORBACHEVA, V.O.; KRASOVA, I.I.; TORAREVA, L.G.; POTEMKINA, Z.I.;

Morphological characteristics of a stabilized capron fiber. Khim. volek. no.3:19-23 64. (MIRA 17:8)

1. Vsesoyuznyy nauchno-issledovatel skiy institut iskusstvennogo volokna.

KULIKOV, K.N.; GORBACHEVA, V.O.

Tensiometer for measuring the tension in moving threads. Khim. volok. no.4263-64 '64. (MIRA 18:4)

1. Vsessoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo volokna.

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000516030003-3"

L 15708-65 EMA(v)/EMT(m)/EMP(j)/T Pc-4/Pe-5 ASD-3/ESD(t)/SSD/AFML/ASD(m)-3 ACCESSION NR: AP4046263 S/0183/64/000/005/0022/0026

AUTHOR: Mikhaylov, N. V.; Gorbacheva, V. O.; Ayzenshtevn, E. M.; Khokhlova, N. S.; Petukhov, B. V.

TITLE: The influence of molecular weight upon the structure and properties of lavsan

SOURCE: Khimicheskiye volokna, no. 5, 1964, 22-26

TOPIC TAGS: synthetic fiber, polyester fiber, polyethylene terephthalate fiber, molecular weight, fiber structure, fiber property, lavsan, polymer crystallization, polymer amorphization, polymer orientation

ABSTRACT: The relation between structure and molecular weight was investigated for lavsan, a polyester fiber, and a fiber from polyethylene terephthalate (PETP) for the purpose of improving the properties of polyester fibers; PETP resembles lavsan at certain stages. Crystallization kinetics, orientation and morphology were determined. Polymers with a 16-30,000 molecular weight and

Card 1/2

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ACCESSION NR: AP4046263

fibers of 17-25,000 molecular weight were investigated; the methods for determining molecular weight and properties are enumerated. Dilatometric curves between 40-140C and density measurements showed that an increase in molecular weight decreased polymer tendency to crystallization. The higher the molecular weight, the broader the interval of the glassforming range (51-96C). Amorphization of PETP increased with increasing molecular weight. So did the coefficient

($\alpha = \frac{\Delta n}{\Delta n_z}$) (double refraction index) for determining the orientation of the isotropic fiber. The same applied to lavsan. Fiber strength paralleled molecular weight; this was obtained at higher temperatures. Data on swelling and dissolution in the out sulfuric and showed fibers with higher molecular weight more resistantly of the anid. Such conditions of structural formation are side good.

errors for obtaining laysan fibers of great strength our goart has Tifig-

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ASSOCIATION: VNIIV

SUBMITTED: 03Aug63

SUB CODE: MT. GC

ENCL: 00

NO REF SOV: 010

OTHER: 004

Card 2/2

GORBACHEVA, V.O.; MIKAHYLOV, N.V.

Differential-thermal analysis of polymers. Vysokom.soed. 7 no.1:28(MIRA 18:5)

32 Ja 165.

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo volokna.

KHAKIMOVA, A.Kh.; KUDRYAVTSEV, G.I.; VASIL'YEVA-SOKOLOVA, Ye.A.; GORBACHEVA, V.O.

Production of cross-linked polyamide fibers. Khim. volok. no.6: (MIRA 18:12)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut iskusstvennogo volokna. Submitted April 27, 1965.

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000516030003-3"

L 37202-66 EWT(m)/EWP(j)/T RM

ACC NR: AP6012418 (A)

SOURCE CODE: UR/0183/65/000/006/0029/0032

65

AUTHOR: Khakimova, A. Kh.; Kudryavtsev, G. I.; Vasil'yeva-Sokolova, Ye.

A.; Gorbacheva, V. O.

ORG: VNIIV

TITLE: Preparation of cross-linked polyamide fibers 5

SOURCE: Khimicheskiye volokna, no. 6, 1965, 29-32

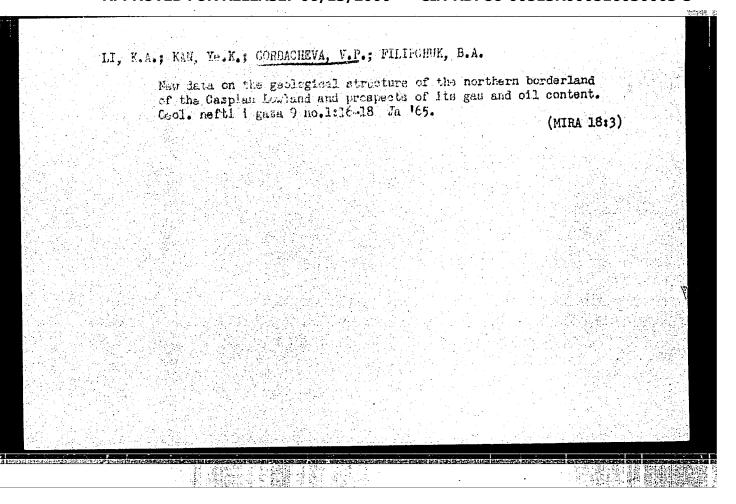
TOPIC TAGS: synthetic fiber, polyamide, polymer structure, IR spectrum, chemical bonding, tensile strength, chemical reaction

ABSTRACT: The process of forming intermolecular bonds in polyamide fibers by reacting with <u>formaldehyde</u> was investigated. Of the acid, neutral and basic catalysts examined, boric acid promoted the best cross-linkages and highest fiber <u>strength</u> Fibers were impregnated with an alcoholic solution of the catalyst, dried and placed in a reactor where they were exposed to a stream of nitrogen and formaldehyde at 135-140°C for 30-120 minutes. The catalyst was then extracted with methanol. Introduction of chemical bonds between the polyamide chains improved deformation properties of the fibers at elevated temperatures,

Card 1/2

UDC: 677.494.675

L 37202-66 ACC NR: AP6012418 reduced solubility, increased zero strength temperature, and doubled heat stability. Data from a chemical method worked out for determining the number of cross-linkages in structured fiber agreed with IR data on the number of substituted amide groups found. A relationship between the number of substituted amide groups found. A relationship between the number of cross-linkages formed and the properties of these fibers was established. As the degree of cross-linking increases, physical phenomena occur which are associated with change in the density of the molecular resident in the fiber. phenomena occur which are associated with change in the density of the molecular packing in the fiber. The authors thank I. O. Novak and Ye. A. Ivanov (LFTI) for conducting IR specroscopic studies on samples of cross-linked fibers. Orig. art. has: 3 tables and 4 figures. SUB CODE: 07 11/ SUBM DATE: 27Apr65/ ORIG REF: 002/ OTH REF: 012 Card 2/2mcP

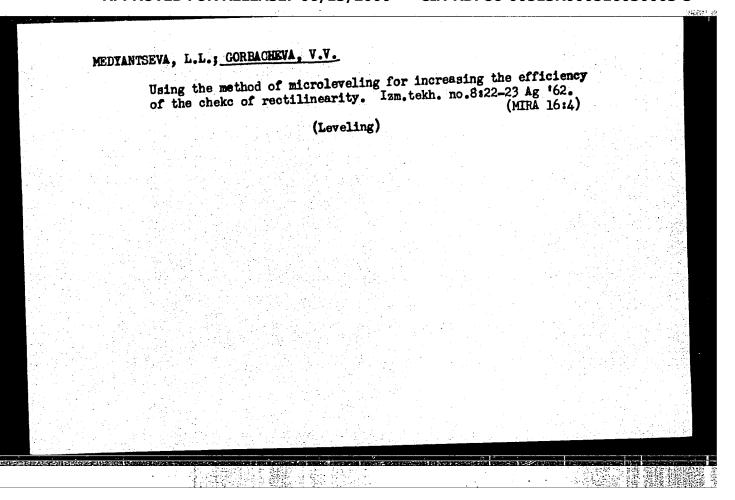


MEDYANTSEVA, L.L.; GORBACHEVA, V.V.

Initial method for controlling large first-grade check rules.

Izm.tekh. no.11:17-18 N '61. (MIRA 14:11)

(Gauges)



	L 44067-56 EWT(m)/EWP(t)/ETT IJP(c) JD ACC NR: AP6030629 (A, N) SOURCE CODE: UR/0413/66/000/016/0125/0125
	INVENTOR: Kakovina, V. G.; Gorbacheva, V. V.; Levina, V. K.
	ORG: none
	TITLE: A method of removing scale from the surface of titanium or its alloys. Class 48, No. 185163/[announced by the Progress Plant (Zavod "Progress")] SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 16, 1966, 125
	TOPIC TAGS: titanium, titanium alloy, titanium electrochemical pickling, titanium alloy electrochemical pickling
	ABSTRACT: This Author Certificate introduces a method for removing scale from the surface of titanium or its alloys by electrolytic pickling in acid solutions containing sodium fluoride. To improve the surface quality, pickling is done in an electrolyte containing (g/1) 400—500 orthophosphoric acid, 30—40 nitric acid, 40—60 sodium fluoride or 180—200 sulphuric acid, 45—50 sodium fluoride, with an initial anodic current density of 1.0—5 a/dm², at a temperature of 40—50C for removing scale which was formed below 700C, or at 70—80C for removing scale which was formed above 700C.
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